

# Molecular Characterization of Heavy Metal Resistant Bacterial Isolates Obtained from Mining Soil in Ikpeshi, South-West, Nigeria

Yakubu P.<sup>1</sup>, Ajayi A. O.<sup>2,\*</sup>

<sup>1</sup>Department of Microbiology, School of Applied Science and Technology, Auchi Polytechnic Auchi, Edo State, Nigeria

<sup>2</sup>Department of Microbiology, Adekunle Ajasin University, Akungba-Akoko, Ondo State, Nigeria; Centre for Infectious Disease Control and Drug Development (CIDCDD), Adekunle Ajasin University, Akungba-Akoko, Ondo State, Nigeria

**Abstract** Mining activities have increased environmental pollution which has consequently resulted in the release of heavy metals into the environment. Bacteria are known to possess several detoxifying mechanisms to withstand the toxic effects of metal species in biotransformation and removal of these metals. The study shows the distribution of the indigenous metal species resistant bacteria in some abandoned mine effluents in Ikpeshi, Akoko-Edo, Nigeria. Soil samples were collected from Goopex quarry (S1), Freedom quarry (S2) Somak quarry (S3) and SC1, SC2, SC3 as their surrounding sites in Ikpeshi, Akoko-Edo, Nigeria. Bacterial isolates were identified by standard microbiological methods. Bacteria tolerance to higher concentration of these metals and their mechanisms of resistance were determined using molecular techniques. A total of 48 metal resistant bacteria species were isolated from three limestone mine tailing and soil in Ikpeshi, Akoko-Edo, Edo State, and using 1mM of Ni, Pb, Zn, Cd, Cr and Co. High resistance to Pb, Ni and Zn were observed in the isolated bacteria strains. Twenty-three of these isolates showed distinct morphological characteristics and the isolates showed tolerance to multiple metals at 5, 9 and 7 mM of Ni, Pb and Zn respectively. Ten isolates showing best growth were selected and their identity confirmed using 16S rRNA. Sequencing of the 16S rRNA gene and phylogenetic analysis of the nucleotide sequences determined from the 16S rRNA gene showed that these isolates belong to the genera *Bacillus*, *Bacteriodes*, *Methanoococcus* and *Deferribacter*, uncultured *Sphingobacterium* and uncultured *Sulfuricurvum*. Screening for the presence of plasmid revealed that the isolates were plasmid mediated. These results indicate that bacteria can be used for bioremediation of heavy metal contaminated site. Hence sampled sites need to be remediated. Measures should be taken to prevent water and wind erosion of the tailing to prevent further dispersal of metal species in the environment.

**Keywords** Bacteria, Environmental pollution, Heavy metals, Metal species, Mine waste

## 1. Introduction

Mining activities have advanced the socio-economic development of national economies around the world; they have impacted negatively on the environment resulting in buildup of pollutants [1]. Incautious use of natural resources coupled with industrialization, mining, and urbanization has resulted in the buildup of pollutants in the environment [2,3]. This leads to great impact on ecosystem functioning because bacteria play a major role maintaining soil fertility and structure [4]. Bacteria are very sensitive and respond quickly to changes in environmental condition; they are therefore considered to be efficient bio-indicators of soil quality [5]. Bacteria in metals contaminated sites are able to overcome

the stress imposed by metal species and thus make them to have a functional role in remediation of the polluted sites [6]. The stress imposed has led to the establishment of various defense mechanisms by the native bacteria community to tolerate the metals [7,8]. Numerous studies have shown that indigenous bacteria isolated from metal species contaminated sites effectively interact with toxic metals via direct and indirect mechanisms. The various resistance ability is a direct response to the particular metal species and consequently a given bacteria may directly and/or indirectly rely on many survival mechanisms [9]. Due to this reason, bacteria are seen as tools for the treatment of metal species polluted sites in biological processes known as bioremediation [10]. They serve as alternative method to physico-chemical processes [11,12]. To effectively use these bacteria in bioremediation, it is compulsory to isolate, cultivate and select the desired strains because bacteria are known to be metabolically

\* Corresponding author:

olajide.ajayi@aau.edu.ng (Ajayi A. O.)

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heterogeneous. Having a better knowledge of the mechanism controlling growth and activity of the bacteria in contaminated sites, their metabolic capabilities and response to environmental changes will also enhance their successful application. Despite large numbers of abandoned limestone mine tailings in Ikpeshi, Akoko Edo, Edo State, there is limited information on metal species resistant bacteria from the various hazardous wastes. Therefore, research on indigenous metal species resistant bacteria associated with mine tailings is important to know the diversity and physiology of metal species resistant bacteria that are thriving under unfavourable metal pollution. Hence, the aim of this research was to isolate metals resistant bacteria associated with limestone mine tailings and soils as well as identify and characterize them for their use as a prospective inoculant in bioremediation of metals polluted sites.

## 2. Methodology

### Soil sampling and analysis

Soil samples were collected from active limestone mines in Ikpeshi, Akoko-Edo Edo State. Control soil samples were collected around the mines. Soil samples were collected in triplicate at depths of 10-30 cm with a soil auger and transported to the laboratory for analysis in sterile plastic bags.

### Preparation of stock solution of metals

All glassware was autoclaved prior to use. The metal salts,  $\text{ZnSO}_4$ ,  $\text{NiCl}_2 \cdot \text{H}_2\text{O}$ ,  $\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$ ,  $\text{CrO}_4$ ,  $\text{CdSO}_4 \cdot 2\text{H}_2\text{O}$  and  $\text{Pb}(\text{NO}_3)_2$  used were of analytical grade purchased from Sigma Aldrich, South Africa. One molar stock solutions of Pb, Ni, Zn, Cd and Co were prepared by dissolving  $\text{ZnSO}_4$ ,  $\text{NiCl}_2 \cdot \text{H}_2\text{O}$ ,  $\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$ ,  $\text{CrO}_4$ ,  $\text{CdSO}_4 \cdot 2\text{H}_2\text{O}$  and  $\text{Pb}(\text{NO}_3)_2$  respectively in milli-Q grade deionized water and later filter sterilized using 0.20  $\mu\text{m}$  pore size membrane filter. The glass wares used were treated in 2N  $\text{HNO}_3$  rinsed several times with distilled water before use to avoid metal contamination [13].

### Isolation of metal resistant bacterial strains

Metal tolerant bacteria were isolated from the samples using the spread plate method. One gram of duplicate composite samples was suspended in 9 mL of saline solution in distilled water and vortexed for 1-2 mins at room temperature. These were serially diluted ( $10^{-1}$  to  $10^{-7}$ ) and aliquots 0.1 ml dilution from  $10^{-4}$ ,  $10^{-5}$  and  $10^{-6}$  were spread with a glass rod over triplicate nutrient agar medium supplemented with 1 mM of  $\text{ZnSO}_4$ ,  $\text{NiCl}_2 \cdot \text{H}_2\text{O}$ ,  $\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$ ,  $\text{CrO}_4$ ,  $\text{CdSO}_4 \cdot 2\text{H}_2\text{O}$  and  $\text{Pb}(\text{NO}_3)_2$  (Van Nostrand *et al.*, 2007). To minimize metal ion complexation, the medium was adjusted to pH 7 using 1.0 N NaOH or 1.0 N HCl. Plates were incubated at 37°C and 25°C for 24 - 48 h. Growth was observed and morphologically distinct colonies were selected and purified in the same medium and growth conditions [14]. All cultures were stored at -80°C in nutrient broth with 20% glycerol for further studies.

### Determination of maximum tolerable metal concentration of the isolates

Maximum tolerable concentration (MTC) for Pb, Cr, Zn, Ni, Cd and Co was determined by agar dilution method [15]. A log-phase culture of the isolates was spot inoculated onto Nutrient agar plates supplemented with increasing concentration of metals. The plates were incubated at 37°C and 25°C for 24 - 48 h and observed for bacterial growth. The highest concentration of the metal at which no bacteria growth was seen was designated as the MTC. All metal salts were added to the Nutrient agar after autoclaving and cooling to 50°C from filter-sterilized stock solutions. The concentrations in mM of the metal species used are stated below:

$\text{Pb}^{2+}$  : 1, 2, 3, 4, 5, 6, 7, 8, 9

$\text{Ni}^{2+}$  : 1, 2, 3, 4, 5

$\text{Zn}^{2+}$  • 1, 2, 3, 4, 5, 6, 7, 8, 9

$\text{Co}^{2+}$  • 0.5, 1, 2

$\text{Cd}^{2+}$  1, 2, 3

$\text{Cr}^{2+}$  • 1, 2, 3, 4, 5, 6, 7, 8, 9, 10

### Determination of multiple metal mixtures tolerance

Multi elemental studies were done using combinations of two metals: Ni+Co, Ni+Pb and Zn+Pb. A log-phase culture of the isolates was inoculated into 10  $\mu\text{L}$  of nutrient broth supplemented with 1mM each of Ni, Pb, Zn and 0.5 mM of Co. The plates were incubated at 37°C and 25°C for 24 h-48 h and observed for bacterial growth by checking the bacterial OD at 600 nm using the UV spectrophotometer. All metal salts were added to the nutrient broth after autoclaving and cooling to 50°C from filter-sterilized stock solutions.

### Characterization and identification of the isolates

Pure culture of bacteria isolates were subjected to various morphological, physiological and biochemical features e.g appearance and colour on agar medium, shape, size of colony, surface elevation, consistency, emulsifiability and gram staining. The result of each test was recorded and the probable identity to the isolates was deduced with references to Bergey's Manual of Determinative bacteriology.

### Physiological characterization

In selecting metal tolerant strains, the physiological conditions necessary for their growth was determined. The physiological conditions that are of great importance to bacteria growth are temperature, pH and salt (NaCl) concentration.

### Salt tolerance

To determine salt tolerance of the isolate, 24 h pure cultures of each bacteria strains were streaked inoculated on Nutrient agar supplemented with 1 mM Pb and 1-10% NaCl which acts as a selective medium and Petri plates were incubated at 25°C and 37°C for 24 - 48 h. Bacteria were marked positive or negative for their ability to grow at different salt concentrations [16].

### pH profile

Hydrogen ion concentration (pH) has a profound effect on

bacteria growth. To determine optimal pH necessary for growth, Nutrient agar supplemented with 1 mM of Pb was used to grow the isolates. It was adjusted to different pH ranging from 3.0 - 9.0 using 1.0 N HCl and NaOH and then inoculated with the pure cultures of the isolates and the plates were incubated at 25°C and 37°C for 24 - 48 h. Growth was measured by the presence or absence of growth on the solidified agar medium.

### Temperature profile

To determine the optimum temperature, the pure isolates were inoculated into Nutrient agar plates supplemented with 1 mM of Pb. Overnight incubation was done at different temperatures; 25, 37, 40 and 50°C. The growth was measured by checking for the presence or absence of growth on the solidified agar plates.

### Molecular characterization of bacterial isolates

#### DNA EXTRACTION

##### Genomic DNA extraction

Using DNA extraction kit from environmental sample (ABM Ontario Canada). Genomic DNA was extracted from the selected bacteria by growing in 10 ml of metal supplemented Nutrient broth. The bacterial cultures were grown for 24 h in a shaking incubator (150 rpm) maintained at 37°C and 1.5 ml of each culture was transferred to sterile Eppendorf tubes which was centrifuged at 1000 rpm for 5 min. Supernatant were discarded and cells were re-suspended in 650 µl of TE buffer. Total genomic DNA was extracted from each bacterial suspension using a ZR soil microbe DNA mini prep TM DNA extraction kit (Zymo Research, USA) according to the manufacturer's protocol.

##### Agarose gel electrophoresis

The presence of the genomic DNA and plasmid DNA was confirmed in a 1.0% (w/v) agarose gel electrophoresis prepared by dissolving 1.0 g of agarose (Bio-Rad, SA) in 100 ml of IX Tris-acetate-ethylenediaminetetraacetate (TAE, pH 8). The mixture was heated in a microwave oven for 3 mins and allowed to cool after which 10 µl of ethidium bromide (Bio-Rad, SA) was added to the molten agar which was poured in a gel casting tray and allowed to solidify. After solidification, the combs were removed and the gel was carefully placed in the electrophoresis tank containing 1 X TAE buffer (40 mM Tris, 20 mM Acetic acid, and 100 mM EDTA pH 8.0). DNA samples were prepared by mixing 5 µl of the genomic DNA with 5 µl of 6X DNA loading dye (Thermo Scientific™) and this was carefully loaded in the pre-formed wells in the gel. A GeneRuler™ DNA Ladder (1 kb) was used to estimate the sizes of the genomic DNA. The electrophoresis was carried out at 100 V, 450 mA for 1 h. Gels were visualized and photographed using a gel documentation system (Gel Doc 2000, Bio-Rad).

##### Polymerase chain reaction (PCR) amplification

The 16S rRNA gene was amplified from genomic DNA obtained from bacterial cultures by PCR using universal

bacterial 16S rRNA primers 27F (5'-AGA GTT TGA TCC TGG CTC AG-3') and 1492R (5'-GGTTAC CTT GTT ACG ACT T-3') previously described by Liu *et al.* (2007). Polymerase chain reaction (PCR) was performed in a total volume of 50 µl containing 25 µl of 2x PCR master mix (0.05 U/µl Taq DNA polymerase, 4 mM MgCl<sub>2</sub> and 0.4 mM dNTPs) (Thermo Scientific™), 2 µl of genomic DNA template, 1.0 µl of each forward and reverse primer and 22 µl of nuclease free water. The amplification reaction mixture was subjected to 30 cycles in a C 1000 thermal cycler (BioRad, USA). The thermal cycling condition used was an initial denaturation, 95°C for 5 min; followed by denaturation at 95°C for 1 min; annealing, 58°C for 30s; extension, 72°C for 1 min and final extension, 72°C for 7 min. The PCR amplicons were analyzed by electrophoresis in 1% (w/v) agarose gel and the sizes of the bands were determined using 1 kb molecular marker. The gel containing ethidium bromide (10 µg/ml) were visualized and photographed using a gel documentation system (Gel Doc 2000, Bio-Rad) to confirm the expected size of the PCR products. PCR products were gel extracted (Zymo Research, Zymoclean™ Gel DNA Recovery Kit).

##### Sequencing reaction

The forward and reverse primers were both used in the sequencing of the purified PCR products. The sequencing was done at Inqaba Biotechnical industrial (Pty) Ltd, Pretoria, South Africa with PRISM™ Ready Reaction Dye Terminator Cycle Sequencing Kit using dideoxy chain termination method and electrophoresed with a model ABI PRISM® 3500XL DNA Sequencer (Applied Biosystems, USA) by following the manufacturer's instructions. Purified sequencing products (Zymo Research, ZR-96 DNA Sequencing Clean-up Kit™) were analyzed using CLC Main Workbench 7.

##### Phylogenetic analysis

##### Sequence similarities and phylogenetic analysis

The chromatograms resulting from sequencing reaction were edited using Chromas Lite version 2.4 software [17]. The resulting nucleotide sequences were then analyzed and edited using Bio Edit Sequence Alignment Editor (Hall and CA, 2004). The consensus 16S rRNA sequences generated were compared with other reference sequences available in the National Center for Biotechnology Information (NCBI) ([www.ncbi.nlm.nih.gov](http://www.ncbi.nlm.nih.gov)) database using the Basic Alignment search tool (BLASTn) [18] and the sequences were deposited in the GenBank. Multiple alignment of the nucleotide sequences were done using Mafft version 7.0 [19]. Phylogenetic and molecular evolutionary analyses were conducted using softwares in MEGA version 6.0 [20]. Evolutionary distance matrix was generated and a phylogenetic tree was drawn by neighbour joining method [21]. The 16S rRNA gene sequences of the selected metal resistant bacterial isolates determined in this study were deposited in NCBI/ EMBL nucleotide sequence GenBank database.

### Plasmid extraction

Plasmid DNA was extracted by the alkaline lysis method as described by Kado [22]. Isolates was inoculated into nutrient broth and incubated for 24hrs at 37°C, after each bacterial culture was suspended in micro centrifuge with lysis buffer. It was heated for 15mins at 70°C and mixed with an equal volume of precooled 100% ethanol. The supernatant was discarded and the pellet was rinse twice with 1ml of 70% ethanol. It was air dried for at least 10mins and finally resuspended in 20µl of TBE (Tris-borate- EDTA) and store in the refrigerator for electrophoresis. The plasmid DNA was run in 1% agarose gel electrophoresis.

## 3. Results

A total of 48 culturable metal species resistant bacteria were isolated from the three sites S1, S2 and S3 under study using ImM concentration each of Pb, Zn, Ni, Co, Cr and Cd. Twenty-eight (23) of these isolates identified by standard microbiological methods clearly tabulated include GS1-*Enterobacter* sp., GS2-*Bacillus subtilis*, GW1-*Sulfuricurvum* sp., GPS2-*Bacillus* sp., GPW2-*Bacillus licheniformis*, BES1-*Zymomonas mobilis*, GPO2-*Enterobacter* sp., GPO4-*Bacillus* sp., BW1-*Zymomonas mobilis*, BW2-*Methanococcus maripaludis*, FW1-*Micrococcus* sp., FS2-*Sphingobacteria bacterium*, FDW1- *Rhodococcus* sp., FDS2-*Bacteroides dorei*, FDS3-*Bacillus* sp. and FD04-*Bacillus subtilis*. other Bacterial isolates encountered were, SW1-*Methanococcus maripaludis*, SW2-*Methanococcus*

*maripaludis*, SS2-*Bacillus subtilis*, SOW1-*Arthrobacter* sp., SOW2-*Bacteroides dorei*, SOW3-*Deferribacter desulfuricans* and SWO4-*Enterobacter* sp. (Table 1). In Table 2, most of the bacterial isolates tested against various metal species including heavy metals shows relatively high resistance at ImM concentration against Chromium ( $\text{Cr}^{2+}$ ), Lead ( $\text{Pb}^{2+}$ ) and Zinc ( $\text{Zn}^{2+}$ ). This ranged from average 6mM to 10mM for these metal species.

Table 3 shows optical density of Bacterial isolates in the multi metal mixtures tolerance test for Nickel and Cobalt (Ni+Co), Nickel and Lead (Ni+Pb), Zinc and Lead (Zn+Pb). The organisms shows significant growth at 48hours, exemplified by the optical density in multi-media cultures within the period of 24hours to 72hours. The growth range continues slowly amongst many of the organisms up to 72hours while few such SS2 -*Bacillus subtilis* and FDS3-*Bacillus* sp. decline for most of this metal mixture. Showing their low metal tolerance.

Physiological characterization of the bacterial isolate was demonstrated in Table 4. Here, most of the bacterial isolates can thrive well at 25°C to 40°C except FD04-*Bacillus subtilis* as well as GS2-*Bacillus subtilis*, GW1 *Sulfuricurvum* sp. that grows slowly and without growth at this range respectively. Furthermore, most of the tested isolates can only grow well at pH 5 to pH 9. Nevertheless, most isolates cannot grow well at salt concentration of 10% and beyond (Table 4). Plates 1 and 2 as well as Figure 1 shows the molecular characteristics and complexity of the bacterial isolates that colonize the ecological niche of the mining sites.

**Table 1.** Total number of metal species resistant bacteria isolated from the three site

Sampling site	Total number of isolates	Number of selected isolates	Designated isolates encountered
S1	16	8	GS1- <i>Enterobacter</i> sp., GS2- <i>Bacillus subtilis</i> , GW1- <i>Sulfuricurvum</i> sp., GPS2- <i>Bacillus</i> sp., GPW2- <i>Bacillus licheniformis</i> , BES1- <i>Zymomonas mobilis</i> , GPO2- <i>Enterobacter</i> sp., GPO4- <i>Bacillus</i> sp., BW1- <i>Zymomonas mobilis</i>
S2	19	8	BW2- <i>Methanococcus maripaludis</i> , FW1- <i>Micrococcus</i> sp., FS2- <i>Sphingobacteria bacterium</i> , FDW1- <i>Rhodococcus</i> sp., FDS2- <i>Bacteroides dorei</i> , FDS3- <i>Bacillus</i> sp., FD04- <i>Bacillus subtilis</i> ,
S3	13	7	SW1- <i>Methanococcus maripaludis</i> , SW2- <i>Methanococcus maripaludis</i> , SS2- <i>Bacillus subtilis</i> , SOW1- <i>Arthrobacter</i> sp., SOW2- <i>Bacteroides dorei</i> , SOW3- <i>Deferribacter desulfuricans</i> , SWO4- <i>Enterobacter</i> sp.
Total number of isolates	48	23	23

**Table 2.** Maximum tolerable concentration of the tested metals against the isolates (ImM Conc.)

Bacterial isolates	Co <sup>2+</sup>	Ni <sup>2+</sup>	Pb <sup>2+</sup>	Cd <sup>2+</sup>	Zn <sup>2+</sup>	Cr <sup>2+</sup>
BW1- <i>Zymomonas mobilis</i>	1	2	4	3	3	8
BW2- <i>Methanococcus maripaludis</i>	1	2	5	1	3	10
SW1- <i>Methanococcus maripaludis</i>	1	2	4	1	4	9
SW2- <i>Methanococcus maripaludis</i>	1	3	7	2	5	7
FW1- <i>Micrococcus</i> sp.	1	3	6	2	5	9
FS2- <i>Sphingobacteria bacterium</i>	1	2	6	2	9	8
SS2- <i>Bacillus subtilis</i>	1	3	4	1	4	8
GS1- <i>Enterobacter</i> sp.	1	3	6	2	7	9
GS2- <i>Bacillus subtilis</i>	1	2	6	2	7	9
GW1- <i>Sulfuricurvum</i> sp.	1	4	6	3	7	9
GPS2- <i>Bacillus</i> sp.	1	4	5	2	7	9
GPW2- <i>Bacillus licheniformis</i>	1	5	7	3	9	10
BES1- <i>Zymomonas mobilis</i>	1	4	6	3	6	7
SOW1- <i>Arthrobacter</i> sp.	1	2	3	1	3	9
SOW2- <i>Bacteroides dorei</i>	1	5	4	2	4	10
SOW3- <i>Deferribacter desulfuricans</i>	1	5	7	3	9	10
FDW1- <i>Rhodococcus</i> sp.	1	4	6	2	5	9
FDS2- <i>Bacteroides dorei</i>	1	5	6	2	4	9
FDS3- <i>Bacillus</i> sp.	1	2	5	2	4	8
GPO2- <i>Enterobacter</i> sp.	1	4	6	2	5	8
FD04 - <i>Bacillus subtilis</i>	1	4	6	2	6	10
SM04- <i>Enterobacter</i> sp.	1	4	6	2	7	9
GPO4- <i>Bacillus</i> sp.	1	4	5	2	7	9

**Table 3.** Optical density of Bacterial isolates in the multi metal mixtures tolerance test

Isolates code	Ni+Co			Ni+Pb			Zn+Pb		
	24hr	48hr	72hr	24hr	48hr	72hr	24hr	48hr	72hr
BW1	0.148	<b>0.229</b>	0.163	0.235	<b>0.248</b>	1.263	0.138	<b>0.226</b>	1.119
BW2	0.005	<b>0.027</b>	0.025	0.996	<b>1.055</b>	1.164	0.547	<b>0.783</b>	0.915
SW1	0.017	<b>0.018</b>	0.017	0.267	<b>0.226</b>	0.156	0.281	<b>0.515</b>	0.713
SW2	0.017	<b>0.064</b>	0.046	0.813	<b>1.319</b>	0.093	0.204	<b>1.170</b>	0.830
FW1	0.079	<b>0.161</b>	0.173	0.287	<b>0.234</b>	0.103	0.369	<b>0.885</b>	0.588
FS2	0.021	<b>0.045</b>	0.050	0.155	<b>0.120</b>	0.014	0.645	<b>1.054</b>	1.194
SS2	0.160	<b>0.216</b>	0.199	0.227	<b>0.170</b>	0.130	0.207	<b>0.377</b>	0.225
GS1	0.037	<b>0.065</b>	0.075	0.223	<b>0.651</b>	0.809	0.350	<b>0.968</b>	0.989
GS2	0.014	<b>0.027</b>	0.031	0.633	<b>0.362</b>	0.636	0.626	<b>0.380</b>	0.460
GW1	0.117	<b>0.998</b>	0.183	0.316	<b>0.412</b>	1.813	0.371	<b>0.761</b>	1.084
GPS2	0.068	<b>0.165</b>	0.112	0.243	<b>0.612</b>	0.912	0.171	<b>0.461</b>	0.953
GPW2	0.015	<b>0.031</b>	0.033	0.751	<b>1.290</b>	1.530	0.750	<b>1.561</b>	1.987
BES1	0.026	<b>0.115</b>	0.120	0.158	<b>1.611</b>	0.723	0.011	<b>0.419</b>	1.591
SOW1	0.010	<b>0.386</b>	0.211	0.424	<b>0.383</b>	0.404	0.776	<b>0.808</b>	0.946
SOW2	0.023	<b>0.027</b>	0.033	0.204	<b>0.482</b>	1.861	0.216	<b>1.099</b>	1.291
SOW3	0.006	<b>0.019</b>	0.024	1.156	<b>1.966</b>	2.310	0.018	<b>1.342</b>	1.822
FDW1	0.216	<b>0.269</b>	0.280	0.241	<b>0.349</b>	0.118	0.114	<b>0.763</b>	0.683
FDS2	0.033	<b>0.030</b>	0.367	0.230	<b>1.215</b>	1.603	0.251	<b>0.750</b>	1.687
FDS3	0.174	<b>0.241</b>	0.247	0.208	<b>0.171</b>	0.241	0.578	<b>0.109</b>	0.085
GPO2	0.050	<b>0.113</b>	0.480	0.221	<b>0.991</b>	0.999	0.856	<b>0.897</b>	0.921
FD04	0.358	<b>0.385</b>	0.151	0.618	<b>0.845</b>	0.661	0.817	<b>1.109</b>	0.211
SM04	0.010	<b>0.016</b>	0.029	0.049	<b>0.174</b>	0.986	0.220	<b>0.690</b>	0.942
GP04	0.168	<b>0.201</b>	0.335	1.107	<b>0.760</b>	1.050	0.116	<b>0.218</b>	0.560

**Table 4.** Physiological characterization of the bacterial isolate

Isolate	Temperature					pH				Salt Tolerance (%)				
	25-30	30-35	35-40	40-45	45-50	3	5	7	9	2	4	6	8	10
BW1	++	++	++	++	++	-	++	++	++	++	++	++	++	-
BW2	++	++	++	++	-	-	++	++	-	++	++	++	++	-
SW1	++	++	++	++	-	-	++	++	++	++	++	++	++	-
SW2	++	++	++	++	++	++	++	+	+	++	++	++	++	++
FW1	++	++	++	++	-	-	++	++	+	++	++	+	+	-
FS2	++	++	++	++	-	-	-	++	++	++	++	++	-	-
SS2	++	++	++	++	-	-	++	++	-	++	+	++	++	-
GS1	++	++	++	++	++	++	++	++	+	++	++	+	-	-
GS2	++	++	++	-	+	+	++	++	+	++	++	++	-	-
GW1	++	++	++	-	++	++	++	+	+	++	++	+	+	-
GPS2	++	++	++	+	+	-	++	+	+	++	++	++	-	-
GPW2	++	++	++	+	++	++	++	+	+	+	++	+	+	++
BES1	++	++	++	++	+	-	++	++	+	++	++	+	-	-
SOW1	++	++	++	+	++	++	++	++	+	++	+	+	++	-
SOW2	++	++	+	+	-	-	++	++	++	++	++	++	+	-
SOW3	++	++	++	+	-	++	++	++	+	++	++	+	+	++
FDW1	++	++	++	-	+	+	++	++	++	++	++	++	-	-
FDS2	++	++	++	-	-	+	++	++	+	+	+	+	++	++
FDS3	++	++	++	-	-	+	++	+	+	++	++	++	-	-
GP02	++	++	++	++	-	+	++	+	+	++	++	+	++	-
FD04	++	++	+	-	-	+	++	+	++	++	++	-	+	-
SM04	++	++	+	++	+	++	++	+	+	++	+	-	+	-
GP04	++	++	+	+	-	-	++	++	+	+	++	-	-	-

Key: + low range or tolerance; ++ high range or tolerance  
 - Negative

**Plate 1.** Fingerprint photograph showing amplicons of approximately 500bp for 16S rRNA gene amplification of the metal resistant bacteria

L	Ladder
L14	BW2
L15	SOW2
L16	FDS2
L17	GW1
L18	BW1
L19	FS2
L20	GPW2
L21	SOW3
L22	SW1
L23	BES1

BACTERIA ISOLATES

- GPW2      *Bacillus licheniformis*
- FS2      *Sphingobacteria bacterium*
- GW1      *Sulfuricurvum* sp.
- FDS2      *Bacteroides dorei*
- SOW2      *Bacteroides dorei*
- BW2      *Methanococcus maripaludis*
- BW1      *Zymomonas mobilis*
- BES1      *Zymomonas mobilis*
- SOW3      *Deferribacter desulfuricans*
- SW2      *Methanococcus maripaludis*

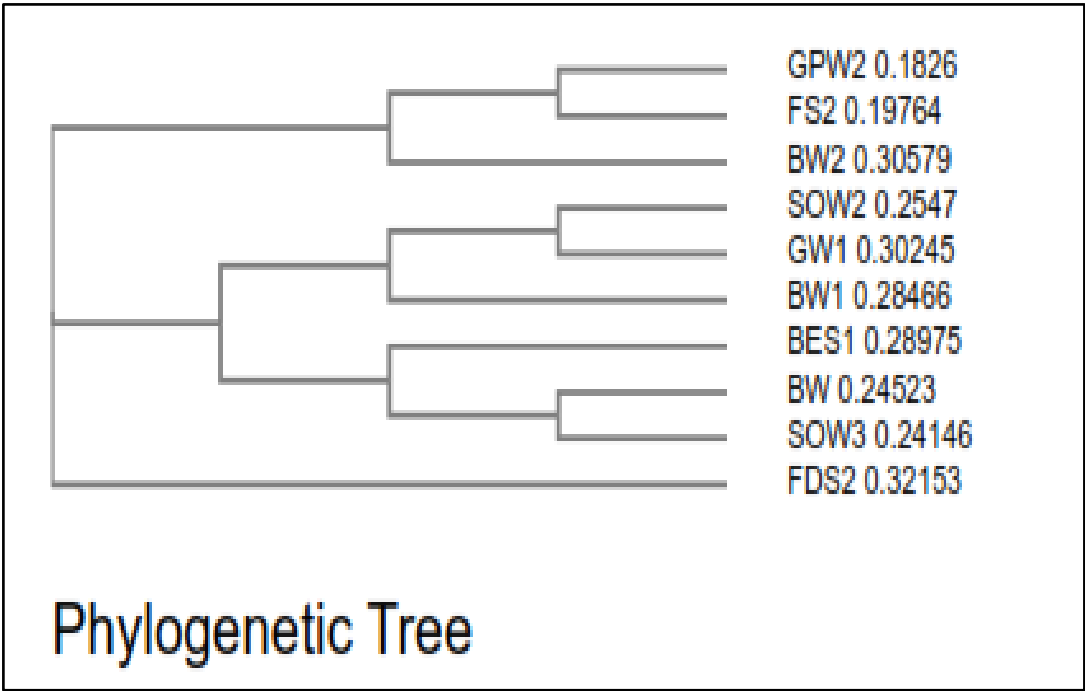


Figure 1. Phylogenetic tree based on partial 16S rRNA gene sequence

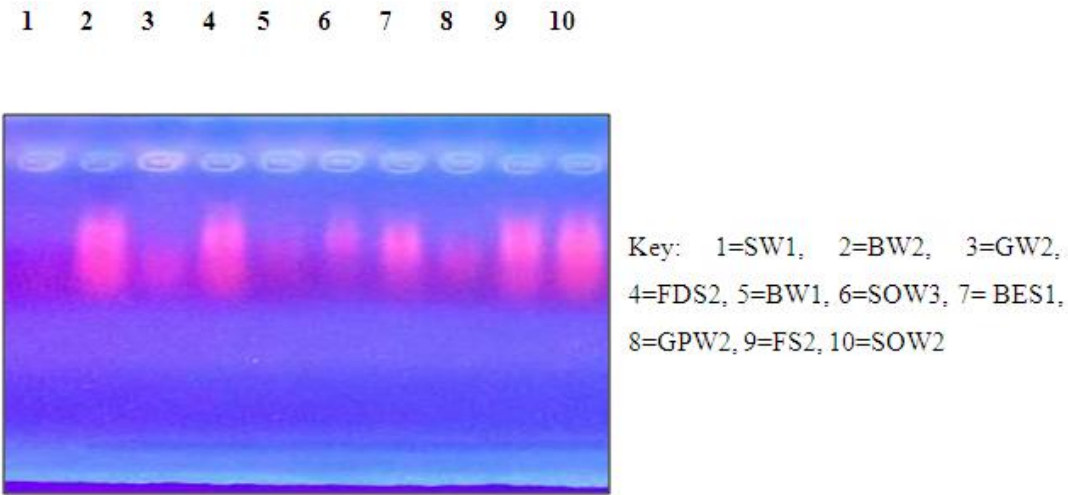


Plate 2. Plasmid profile

## 4. Discussions

A total of 48 culturable metal resistant bacteria were isolated from the three sites under study using ImM concentration each of Pb, Zn, Ni, Co, Cr and Cd. 23 of these isolates shows distinct colony characteristics like size, pigmentation, shape, elevation, and margin surface (Table 1). All the bacterial isolates tested could not grow beyond 1 mM of Co, but they were able to tolerate varying concentrations of Ni, Pb, Zn, Cd and Cr as shown in (Table 2). Cadmium tolerance range between 1-3 mM and five isolates; GPW2, BES1, BW1, SOW3 and GW1 could tolerate up to 3 mM concentration of this metal. Bacterial tolerance to  $\text{Ni}^{2+}$  range between 1-5 mM, isolates SOW2, FDS2, GPW2 and SOW3 shows highest tolerance of 5 mM to this metal. Tolerance to Pb also ranges between 1-7 mM for the isolates; SW2, GPW2 and SOW3 show the highest tolerance of 7 mM. Similarly, tolerance to Zn was also found to range between 1-9 mM and three isolates GPW2, SOW3 and FS2 were able to tolerate up to 9 mM concentration of this metal. Chromium was also well tolerated by the isolates; high tolerance of 7-10 mM was recorded. Isolates GPW2, SOW2, BW2 and SOW3 shows highest tolerance of 10 mM. Bacterial isolates GPW2 and SOW2 were observed to tolerate the highest concentration of 5 mM Ni, 7mM Pb, 9 mM Zn and 10 mM  $\text{Cr}^{2+}$  of the metal species tested. The high level of resistance and the widespread tolerance shown by the bacterial isolates against multiple metals could be attributed to the elevated concentrations of the metal species recorded in the different samples where the bacteria were isolated from. The differences in metal tolerance by the bacterial isolates could be attributed to many factors such as the medium strength, presence of negatively charged ions like chloride, organic constituents, and nature of the medium which determines availability of the metals to the bacteria (Kannan and Krishnamoorthy, 2006). The present study also shows that the bacterial isolates have multiple metal resistances (Table 3). The isolates show different tolerance to the multiple mixtures of Imetals tested. It was observed that Ni+Co combination was generally toxic to the bacterial isolates as observed in their reduced growth rate (Table 3). The bacterial isolates were able to tolerate the combination of Ni+ Pb and Zn+Pb better than Ni+Co. Out of the 23 isolates, ten isolates shows better growth in the presence of the multiple metal mixtures but the isolate GPW2 and SOW3 shows the best growth. All the isolates were further characterized for their physiological and biochemical characteristics to determine the probable identity of bacteria that could tolerate multiple metals in the mine tailings.

The effect of temperature, pH and salt on bacteria growth in the presence of metal is shown in Table 4 The results shows that the optimum temperature for maximum growth by all the bacterial isolates was between 30°C-35°C, although majority of the isolates were also able to grow between 35°C-40°C. Decline in growth was observed when the temperature was raised to 40°C which could be due to the

decrease in metabolic activity as a result of increase in temperature above the optimum value the bacteria can withstand. The observation corroborate with the studies of Adamo *et al.*, [23] and Bajkic *et al.*, [24], who reported that High temperature of growth medium affects the arrangement and stability of bacterial cell wall, ionized chemical moieties and also leads to disruption of the enzymatic activities which consequently slow down the metabolic activities of the bacteria. Similarly, low temperature also reduced bacterial growth because enzymes are inactivated at this temperature which decreases the rate of metabolism [25,26,27].

The result of the effect of pH on bacterial growth indicated that optimum pH range for the bacterial growth is 5 and 7 while a decline in growth was observed at pH 3 and 9. The effect of salt on bacterial growth shows that increasing concentration of salt in the growth medium reduces the bacteria growth rate, maximum growth was observed at 2% followed by 4%. The result obtained is in accordance with the report of Fashola *et al.* [28] who showed that the bacteria are able to produce osmolytes like sugar and amino acids to protect themselves against the hypertonic environment created by the salt. At higher concentration of 6% -8%, growth was greatly reduced, some of the isolates could not withstand the high osmotic gradient created by salt again while at 10% only isolates SW2, GPW2, SOW3, and FDS2 could tolerate the high salt concentration in the growth medium (Table 4). The observation correlates with the study of Fashola *et al.* [28], who reported that effect of salt on bacterial growth shows that increasing concentration of salt in the growth medium reduces the bacteria growth.

Ten isolates: GPW2, FS2, BW2, SOW2, GW1, BW1, BES1, SW2, SOW3 and FDS2 showed high metal tolerance. The amplification of the 16S rRNA genes from the genomic DNA yielded a 1.5 kb fragment as shown in (Plate 1). The BLAST query grouped the bacterial isolates into *Bacillus*, *Bacteriodes*, *Methanococcus*, *Deferribacter*, *Sphingobacterium* and *Sulfuricurvum* sp.as shown in Fig 1. Exposure to metals has been reported to usually lead to tolerance in population of Gram positive and Gram negative bacteria present in contaminated sites. [29] as observed in this study. None of the isolates shows resistance to the metal species on the chromosomal DNA confirming the earlier claim that resistance to metal are majorly located on plasmid (Plate 2). These bacteria are known to apply various types of resistance mechanisms which could be plasmid or chromosomally encoded but these resistance has been mostly reported to be plasmid mediated [28,30,31]. In a study conducted by Raja and Selvam [14], resistance to Pb and Ni by *Pseudomonas aeruginosa* isolated from metal polluted waste water was found to be plasmid mediated. Similarly, Piotrowska-Seget *et al.* [32] also reported bacterial resistance to Zn in their study to be plasmid mediated. The presence of metal resistance determinants on plasmids has suggested that these genes may be transferred to divergent bacteria through horizontal gene transfer [33].

## 5. Conclusions

This research invariably establishes some novel characteristics of isolates strains which had high level of adaptability to the mines environmental with biodegradable ability to degrade some heavy metals that are of high significance. This study is the first to report a heavy metal tolerance capacity of bacteria isolated from the soil of Ikpeshi, Akoko Edo, and mining areas. Studies undertaken to examine the identification and characteristics of environmental samples revealed the true diversity of microorganisms and their unique functionality which arise from their biological system that produce enzymes to make them tolerate or adapt to their environments. The use of molecular techniques adds more precision and accuracy to the phylogenetic identification and the true reflection of microbial diversity. *Bacillus* can be used for knowledge of the active microorganisms in the mines and is important for recovering metals and the development of optimal *in situ* bioremediation strategies. Many organisms encountered in this study shows significant growth at 48hours, exemplified by the optical density in multi-media cultures within the period of 24hours to 72hours. The growth range continues slowly amongst many of the organisms up to 72hours while few such SS2 -*Bacillus subtilis* and FDS3-*Bacillus* sp. decline for most of this metal mixture. Showing their low metal tolerance. That means, other bacterial isolates studied in this context can be used for bioremediation purposes which can safeguard the ecology and health of people in this are and human health generally.

## 6. Recommendations

Measures should be taking to prevent water and wind erosion of the tailings as this will lead to further dispersal of metal species in the environment and the sampled sites need to be remediated urgently. More funding and support should also be provided by funding agencies, research development bodies and government to encourage implementation of bioremediation technology on a large scale as this hold promising future for the treatment of the enormous mine wastes generated by the country. Complementary to this further research can be done on the use of genetically modified organisms for this bioremediation purpose as well as the possibility of converting this waste sources for some valuable energy purpose in bid of making resource from the waste. Good database should be formulated for research development and health management system in this regard.

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