

Analysis and Simulation of MEMS Based Multiple Gases Detection System Using COMSOL

K. K. Singh¹, Hardeep Kumar^{1,*}, Anuj Kumar², Gaurav Chaudhary¹

¹Department of Electrical and Electronics Engineering, BITS Pilani, Dubai Campus, Dubai, U.A.E.

²Department of Mechanical Engineering, BITS Pilani, Dubai Campus, Dubai, U.A.E.

Abstract In the recent past, there have been a number of researches on integrated gas sensors tailored for various applications such as disease detection, pollution monitoring and many more. Most of the previous designs were based on the “micro-array” type. However, this could significantly consume huge space on the substrate. With an interest of providing a cost effective solution, a novel micro/nano engineered gas sensing chamber has been designed. In this paper, the mathematical model of the gas detection mechanism for the evaluation of the system performance has been discussed. The model has been optimized for high sensitivity operation and the optimal velocity of the gas flow in the chamber has been found to be $2e-5$ m/s for NO_2 and H_2 in air of concentrations 100 ppm and 2000 ppm respectively, at 50°C .

Keywords MEMS, Gas Sensor, MHP, Gas Chamber, COMSOL

1. Introduction

The idea of “gas sensors” has been conceived in the early 18th century. Catalytic combustion (LEL) sensor is the first type of gas sensor developed [1]. The mechanism of LEL sensor is based on the fact that the oxidizing gases increase the resistance of the hot wire filament. In 1925, Dr. Uzumi Doi has invented a new interference based technology for the gas detection at the Institute of Physical and Chemical Research, Japan [2]. This was very useful in the monitoring of the explosion of methane vapours from the oil tankers. This follows the method of light diffraction in air. In early 80s, MDA pioneered toxic detection using paper tape technologies [3].

The concept of miniaturization in the field of “gas sensors” has originated in the year 2000 through the invention of watch type portable gas monitor. Since this invention, many researchers in this field focused extensively in bringing out miniaturized gas sensor products [4-6]. Micro-electromechanical System (MEMS) technology has aided these researchers in their objective. In most of the initial attempts, there were inextricable issues due to the cross-sensitivity towards the background gases. Hence, the signal to noise ratio (SNR) was poor. Later researches were concentrated on an effective sensing of gases with maximal SNR [7].

With the blooming of chemical industries, oil refineries, etc. in this modern world, there is a potential danger due to

multiple poisonous gases exiting out. This calls for an integrated monitoring of the gases. The similar situation is associated with the chemical/biological research laboratories [8-10]. Further, an integrated gas sensor is an unavoidable component for the world’s future smart infrastructure initiative.

Gases are linked to all forms of life and their odors tremendously influence the perception of our environment. The human nose serves as the highly advanced sensing instrument that is able to differentiate between hundreds of smells. But it cannot detect absolute gas concentrations or sense odorless gas. Various sensing technologies are often used in fire alarms and smoke detectors including heat detectors, photoelectric detectors (which detect smoke), and ionization detectors (which are activated by the airborne particles of combustion).

In addition to the above applications, an integrated gas sensor has also influenced the healthcare industry in a positive way. In the healthcare sector, Gas sensors are referred by its popular jargons: “an electronic nose” or “breath analyzer”. They examine odors from the body or from the human breath and serve to identify possible problems. “For example, odors in the breath can be indicative of gastrointestinal problems, infections, infected wounds and tissues emit distinctive odors that can be used for diagnosis, and odors coming from body fluids can indicate liver and bladder problems [11].

The previous designs of MEMS based integrated gas sensors were based on “micro-array” type [3]. However, this consumes a lot of space on the substrate. This way, the cost of manufacturing/fabrication increases. In this paper, we present an alternative solution for the cost-effective production of integrated gas sensors. The design employed

* Corresponding author:

hardeep@dubai.bits-pilani.ac.in (Hardeep Kumar)

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is based on tube geometry with nano-engineered pores [15, 16].

The paper is structured in 3 sub-sections. The first sub-section deals with the description of physical architecture of the gas sensor system. The second sub-section deals with the mathematical modelling related to the detection process. The last sub-section deals with the optimized solution of the mathematical model [12-14].

2. Physical Architecture of Gas Sensor

The gas detection chamber is tube shaped with an alternative assembly of electrically conductive and non-conductive micro-rings as shown in Figure 1. The electrically conductive micro-discs are the gas sensitive elements in the chamber. These are pre-heated to the requisite temperatures for optimal detection. The heating is accomplished using Micro-Hot Plates. The changes in the resistance of conductive micro-discs provide information about the concentration of gas flowing through the chamber. For the proof of concept, the suspension of NO₂ and H₂ gases in air is considered and the ‘m’ and ‘n’ electrically conductive micro-discs respectively, are tailored for their detection. The non-conductive micro-discs are used for the electrical isolation of two conductive micro-discs.

3. Mathematical Modeling and Numerical Simulation

3.1. Micro Hotplate Design

Micro hotplate (MHP) is one of the components in sensing system. It is placed adjacent to the sensing layer. There is a need to elevate the temperature of the sensing layer for the capture of gas analytes. This can be made possible by using micro hotplate. A best fitted circular type porous MHP has been designed.

The electrical field distribution, E , across the heater structure is determined by the equation,

$$E = -\nabla V \tag{1}$$

Where, ∇V is the potential difference across the micro heater. The boundary condition for this equation is defined by the applied constant potential at both heater contacts,

Which are defined at the base contacts of the anchors of the beam. The physics of Joule’ heating can be modeled by the equation 2.

$$\rho C_p \nabla T = \nabla \cdot (k \nabla T) + Q_j \tag{2}$$

Where, ρ is the electrical resistivity of heater material, C_p is the heat capacity at constant pressure of heater material, k is the thermal conductivity of the heater material and ∇T is the temperature difference across the heater material. The temperature distribution in the pore micro-hotplate is depicted in Figure. 2.

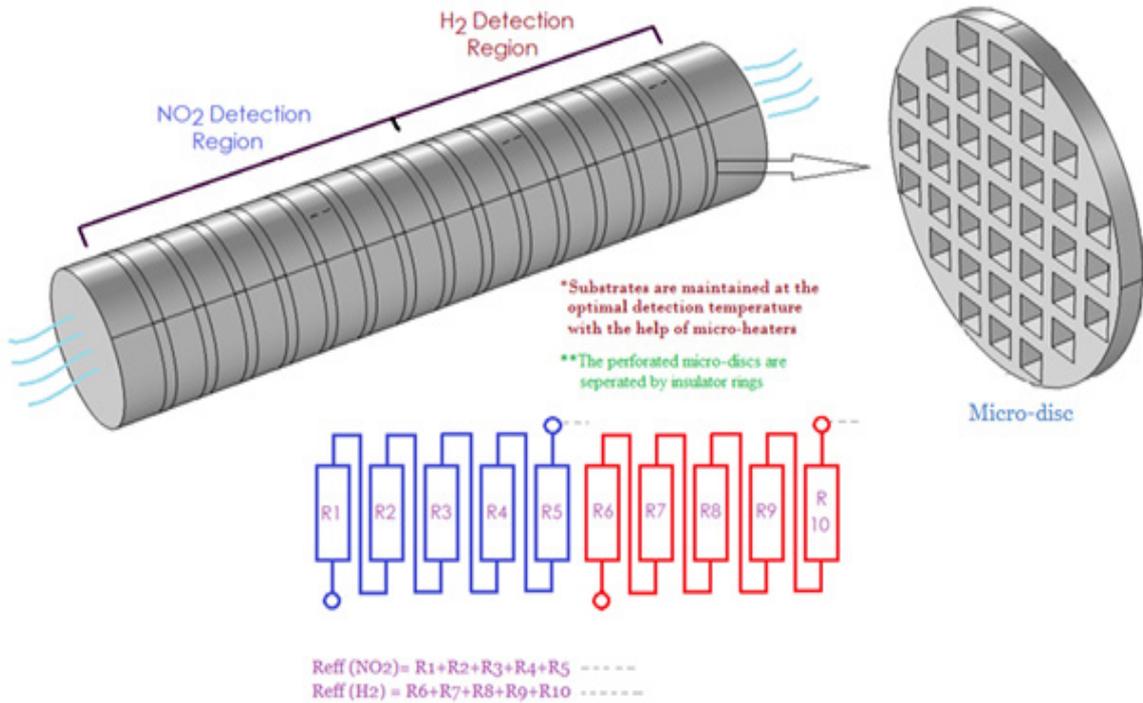


Figure 1. Architecture of the Gas sensor system

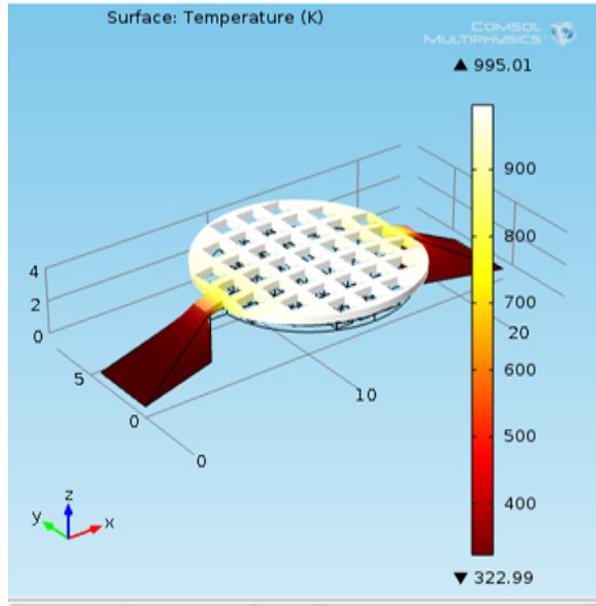


Figure 2. Temperature distribution in pore micro hotplate

3.2. Gas Capture

The laminar flow condition is assumed for the flow on this device and therefore Navier-Stokes equation is used to model the laminar flow inside the micro-fluidic channel.

$$\rho(u.\nabla)u = \nabla.[-pI + \mu(\nabla u + (\nabla u)^T) - 2/3 \mu(\nabla.u)I] \quad (3)$$

$$\nabla.(\rho u) = 0 \quad (4)$$

Where, p is the pressure at the inlet, ρ is the density of the air; μ is the dynamic viscosity of the air. Laminar flow boundary conditions are as follows: At the opening of the chamber i.e., $u = 2 \times 10^{-5}$ m/s at the rear end of the chamber, $p = 1$ atm and at the walls, $u = 0$ m/s.

The mass transport of the analyte through the sensing micro-disc is modeled by the following mass transport equations.

$$\nabla \cdot (-D_i \nabla c_i) + u \cdot \nabla c_i = 0 \quad (5)$$

$$N_i = -D_i \nabla c_i + u c_i \quad (6)$$

Where, D_i is the diffusion coefficient of analyte in air, c_i is the analyte concentration, u is the velocity field and N_i is the total flux of analyte transport. The metal oxide- micro-disc reacts with the analyte.

The concentration at the outlet,

$$-n \cdot D_i \nabla c_i = 0 \quad (7)$$

The following condition is applied for the coating.

$$\nabla \cdot (-D_i \nabla c_i) + u \cdot \nabla c_i = N \quad (8)$$

Where, N is the reaction rate in mol/m³/s. The ‘no flux’ condition is applied to the walls, i.e.

$$-n \cdot N_i = 0 \quad (9)$$

The mathematical model was implemented in COMSOL Multiphysics. The governing equations and the boundary conditions are provided in the model. Size-adaptive tetrahedral meshing is used to solve the equations.

3.3. Optimized Solution

The PDEs 1, 3, 4 and 9 were optimized to maximize the capture of analyte in the sensing micro-disc. At 50°C, the capturing of NO₂ and H₂ in air of concentrations 100 ppm and 2000 ppm respectively is maximum at the air flow velocity of 20 μm/s, as shown in Figure 3. The number of micro-discs required to detect the concentration of NO₂ and H₂ in air is 4.

The concentration profile of NO₂ and H₂ gases are shown in Figure 4 and 5. The 100 ppm of NO₂ gas suspended in air is sent in at the inlet. As shown in the Figure 5, the concentration of gas gradually decreases as it goes through the chamber. This is due to its capture in the conductive Silicon micro-discs. After getting diffused through 4 micro-discs, the NO₂ concentrations dip to almost 0.

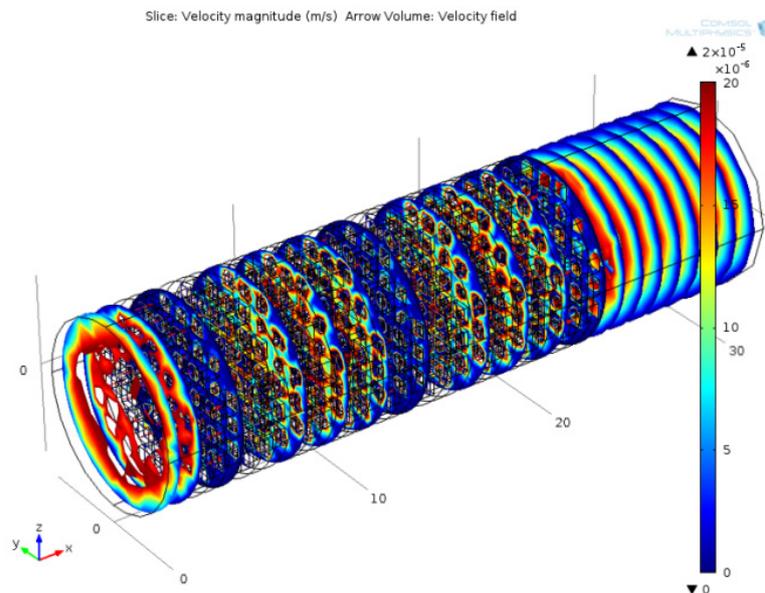


Figure 3. Velocity profile of the air flow in the chamber

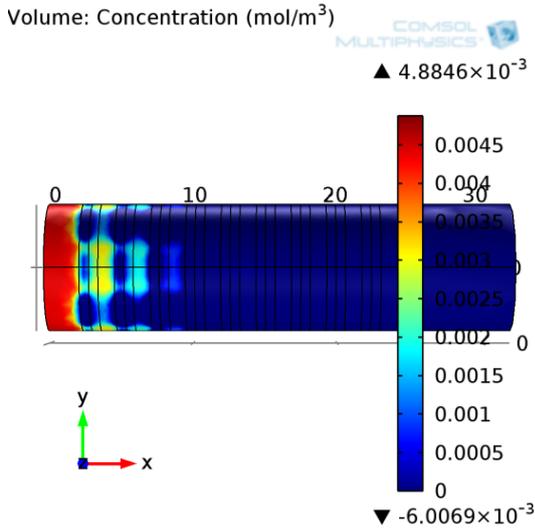


Figure 4. Concentration profile of NO₂ gas in the Chamber

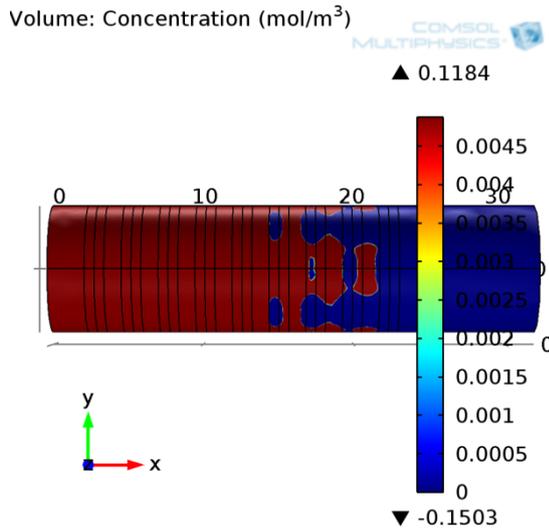


Figure 5. Concentration profile of H₂ gas in the Chamber

Similar mechanism takes place in the case of H₂ gas. In this case, H₂ gas of 2000 ppm concentration is captured by Pd micro-disc. The relative change in the resistance of the micro-discs is directly relative to the concentration of analyte captured. The ‘m’ and ‘n’ sensing micro-discs are connected in series respectively. This way, the measurement of the overall relative change in the resistance will help us in finding out the concentration of different analytes suspended in the air flowing through the chamber. The relative change in resistance has been plotted as shown in Figure 6 and 7.

In the Figures 6 and 7, the bluish blocks represent the hollow region and the reddish blocks represent the micro hotplate. The relative change in the resistance of the Si and Pd micro-disc has been plotted in the figures 6 and 7 respectively. These micro-discs should be electrically connected in series separately to get the overall system response for each analyte.

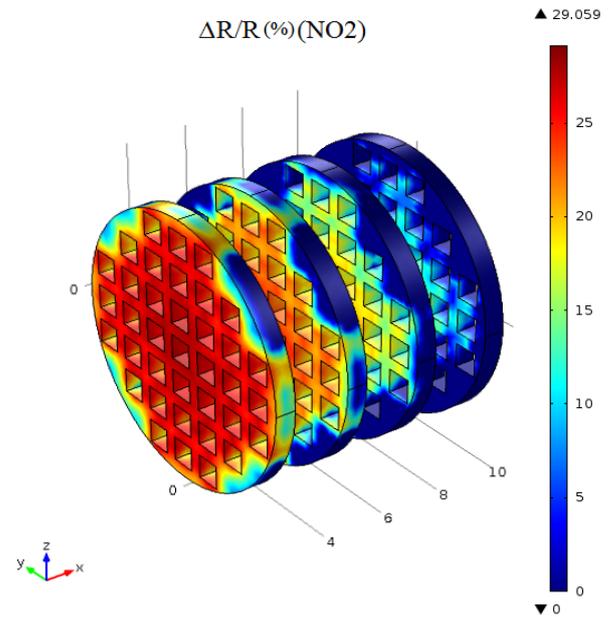


Figure 6. Relative change in resistance of Si micro-disc

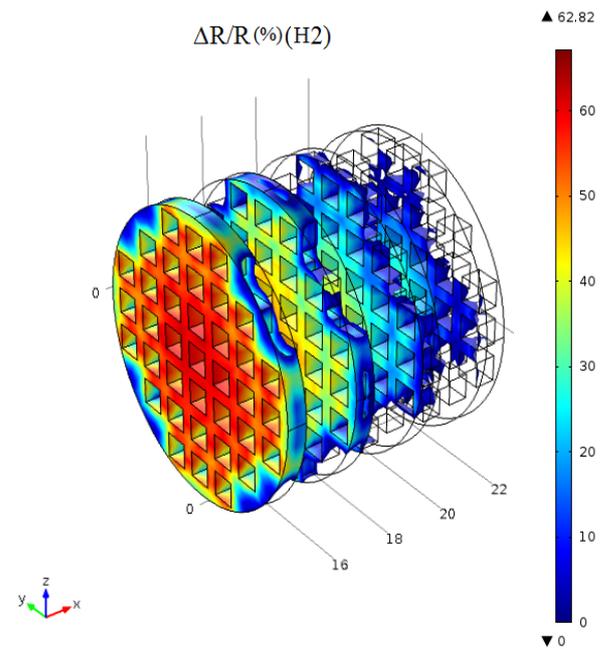


Figure 7. Relative change in resistance of Pd micro-disc

4. Conclusions

A tube shaped micro gas sensor system has been designed for the selective capture of NO₂ and H₂ gases from the air. The results show that the optimal velocity of the air flow in the chamber is 20 μm/s. Further, a micro hotplate has been design to heat the micro-discs to 50°C. The resistance change due to chemical absorption of the gases can be quantified using signal processing circuits, thereby paving way for the detection of gas analytes.

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